THE . MERCK INDEX.

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THE MERCK INDEX

AN ENCYCLOPEDIA OF CHEMICALS, DRUGS, AND BIOLOGICALS

ELEVENTH EDITION

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Published by
MERCK & CO., INC.
RAHWAY, N.J., U.S.A.

1989

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Library of Congress Catalog Card Number 89-60001 ISBN Number 911910-28-X

Printed in the U.S.A

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Isoln from mustard seeds: Goering, U.S. pat. 2,987,448 (1961 to Oil Seed Prod.).

a-Form, monohydrate, needles, mp 159°. $[\alpha]_0^{\infty}$ ÷98.0° → 50.0° (water). Soluble in water, slightly sol in hot alcohol. Practically insol in ether.

ractically insol in editer. β-Form, mp 166°. [a]₀ +27° → +55.6° (water). Phenylhydrazone, mp 141°.

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4243. Galanga. Galangal; colic root; East India root; Chinese ginger. Dried rhizome of Alpinia officinarum Hance. Zingiberaceae. Habit. China. Constit. Volatile oil, resin, kaempferid, galangin, dioxyflavanol, galangol.

4244. Galangin. 3,5,7-Trihydroxy-2-phenyl-4H-1-bentopyran-4-one; 3,5,7-trihydroxyflavone; norizalpinin. Cl₃H₁₀O₃: mol wt 270.25. C 66.67%, H 3.73%, O 29.60%. Isoln
from galanga root. Alpininia officinarum, Hance and characterization: E. Jahns, Ber. 14, 2807 (1881). Prepn: T. Heap.
R. Robinson, J. Chem. Soc. 129, 2336 (1926); J. J. Chavan,
R. Robinson, ibid. 1933, 368. Mutagenicity studies: J. T.
MacGregor, L. Jurd, Mutat. Res. 54, 297 (1978); J. P.
Brown, P. S. Dietrich, ibid. 66, 223 (1979).

Yellowish needles from ethanol, mp 214-215°. Moderately sol in ethanol, ether; insol in water. Very sol in chloro-

4245. Galanthamine. 4a,5,9,10,11,12-Hexahydro-3-methoxy-11-methyl-6H-benzofuro[3a,3,2-ef][2]benzazepin-6-al; galantamine; lycoremine; Jilkon. C₁₇H₁₁NO₃; mol wt 287.35. C 71.05%, H 7.37%, N 4.87%. O 16.70%. From Caucasian snowdrops. Galanthus woronowii Vel., Amaryllidaceae: Proskurnina, Yakovleva, J. Gen. Chem. 22, 1899 (1952); from Narcissus spp: Boit et al., Ber. 90, 725, 2197 (1957). Structure work: Kobayashi et al., Chem. & Ind. (London) 1956, 177. Synthesis and stereochemistry: Barton, Kirby. Proc. Chem. Soc. 1960, 392; J. Chem. Soc. 1962, 806; Williams. Rogers. Proc. Chem. Soc. 1964, 357. Alternate total synthesis: Kametani et al., J. Chem. Soc. (C) 1971, 1043. Asymmetric synthesis of (+)- and (-)-forms from L-tyrosine: K. Shimizu et al., Heterocycles 8, 277 (1977). Biosynthesis studies: D. H. R. Barton et al., J. Chem. Soc. 1963, 4545; W. Döbke, Heterocycles 6, 551 (1977). 4245. Galanthamine. 4a,5,9,10,11,12-Hexahydro-3-4545; W. Döbke, Heterocycles 6, 551 (1977).

Crystals from benzene, mp 126-127. [α] 8 0 -118.8° (c = 1.378 in ethanol). Monoacidic base. Fairly sol in hot waters freely sol in alcohol, acetone, chloroform. Less sol in ben-

zene, ether. Hydrochloride, C₁₇H₂₁NO₂.HCl, crystals from water, dec 256-257. Sparingly sol in cold, more sol in hot water. Very sparingly sol in alcohol, acetone.

Hydrobromide, C11H11BrNO3. Nivalin. Crystals from

water. dec 246-247°. [α] $_D^{80}$ =93.1° (c = 0.1015 in 15 ml H,O). LD $_{50}$ i.v. orally in mice: 8.0, 18.7 mg/kg, Umarova et al., C.4. 66, 53993v (1967). THERAP CAT: Cholinesterase inhibitor.

4246. Galegine. (3-Methyl-2-butenyl)guanidine; N-3.3-dimethylallylguanidine; isoamyleneguanidine. C₆H₁₃N₃; mol wt 127.18. C 56.66%, H 10.30%, N 33.04%. Isoprenoid guanidine deriv from seeds of Galega officinalis L., Leguminosae: Tanret. Compt. Rend. 158, 1182, 1426 (1914): 159, 108 (1914): Markovic, Dittertovà. Chem. Zvesti 9, 576 (1955), C.A. 50, 8137d (1956). Structure: Barger, White, Biochem. J. 17, 827 (1923). Synthesis: Späth, Spitzy, Ber. 88, 2273 (1925); Babor, Jezo, Chem. Zvesti 8, 18 (1954), C.A. 49, 7495f (1955). Metabolic effects: G. Weitzel et al., Z. Physiol. Chem. 353, 535 (1972). Effects on mitochondria: B. Lotina et al., Arch. Biochem. Biophys. 159, 520 (1973). Biosynthetic study: J. Steiniger, G. Reuter, Biochem. Physiol. Pflanz. 166, 275 (1974). Review: Braun, J. Chem. Ed. 8, 2175 (1931). 4246. Galegine. (3-Methyl-2-butenyl)guanidine; N-3.3-

Hygroscopic, bitter crystals. mp 60-65°. Freely sol in water or alcohol, slightly in ether. Keep well closed.

4247. Galipine. 2-[2-(3,4-Dimethoxyphenyl)ethyl]-4-methoxyquinoline. C₂₀H₃₁NO₃; mol wt 323.38. C 74.28%, H 6.55%, N 4.33%, O 14.84%. From Angostura bark (Cusparia trifoliata Engl., Rutaccae): Körner, Böhringer, Gazz. Chim. Ital. 13, 363 (1883); Tröger, Kroseberg, Arch. Pharm. 250, 1494 (1912). Synthesis: Späth, Eberstaller, Ber. 57, 1687 (1924); Späth, Pikl, Ber. 62, 2244 (1929); Schläger, Leeb, Monatsh. 81, 714 (1950).

$$\underbrace{\mathsf{CH}_2\mathsf{CH}_2}_{\mathsf{OCH}_3} \underbrace{\mathsf{CH}_3}_{\mathsf{OCH}_3}$$

Prismatic needles from alc. mp 116°. Soluble in alcohol, benzene, chloroform, ether; slightly sol in water, petr ether. The salts are more sol than those of cusparine. Hydrochloride tetrahydrate, C₁₀H₂₁NO₃.HCl.4H₂O, plates, become anhydr at 100°, mp 165°. Methiodide, C₂₀H₂₁NO₃.CH₃I, yellow needles, mp 146°.

4248. Gallacetophenone. 1-(2,3,4-Trihydroxyphenyl)eth-4248. Gallacetophenone. 1-(2,3,4-Trihydroxyphenyl)ethanone; 2',3',4'-trihydroxyacetophenone; Alizarine yellow C; C.I. 57000. C₄H₄O₄; mol wt 168.14. C 57.14%, H 4.79%, O 38.06%. Prepn: Hart, Woodruff, J. Am. Chem. Soc. 58, 1957 (1936); Campbell. Coppinger, U.S. pat. 2,686,123 (1954 to U.S. Secy. Agr.); Knowles, U.S. pat. 2,763,691 (1956 to Kodak); Price, Israelstam J. Org. Chem. 29, 2800 (1964)

White to brownish-gray, cryst powder, mp 173°, uv max (methanol): 237, 296 nm (ε 8560, 12,500). Sol in 600 parts cold water, more in hot water; sol in alcohol, ether, soln of sodium acetate.

USE: Antiseptic.

4249. Gallamine Triethiodide. 2,2',2"-[1,2,3-Benzenetriyltris(oxy) tris(N, N, N-triethylethanaminium) triiodide; -phenenyltris(oxyethylene)]tris[triethylammonium triiodidel; 1.2.3-tris(2-triethylammonium ethoxy)benzene triiodide: 1.2.3-tris(2-diethylaminoethoxy)benzene tris(ethyl iodide); tri(8-diethylaminoethoxy)-1.2.3-benzene triiodo-

ny: acetone 43.5, benzene 28.9, CHCl₃ 24.0, ether 20.8, chanol 6.4. Insol in water. LD₅₀ in male, female rats tags kgl; 88, 91 orally (Gaines).

Human Toxicity: Poisoning may occur by ingestion, inhalation, or percutaneous absorption. Acure: Dizziness, tags and poison and the poison of the p halation, or personance association. Acute: Dizziness, headache, nausea, vomiting, diarrhea, tremors, weakness, conculsions, dyspnea, cyanosis, circulatory collapse. Chron-the Hepatic damage has occurred in expt animals. Topical acute level sensitivity regulatory. Hepatic damage has occurred in expit animals. Topical use may cause local sensitivity reactions. Vapors may irritate eyes, nose, throat. See: Clinical Toxicology of Commercial Products. R. E. Gosselin et al., Eds. (Williams & Williams, Baltimore, 5th ed., 1984) Section III, pp 239-241, Lindane and other hexachlorocyclohexane isomers may reachibly be anticipated to be carrinogene. Fourth Apparent casionably be anticipated to be carcinogens: Fourth Annual Report on Carcinogens (NTP 85-002, 1985) p 123.

USE: Insecticide. IHERAP CAT: Pediculicide; scabicide. HIFRAP CAT (VET): Ectoparasiticide.

5380, Lindlar Catalyst, Pd-Pb-CaCO₂, Prepn: Lindlar, Relv. Chim. Acta 35, 446 (1952); Lindlar, Dubuis cited by Fieser, Fieser, Reagents for Organic Synthesis (New York, 1967) p 566.

In selective hydrogenation of triple bonds to cisdouble bonds.

5381. Lineatin. [1R-(1\alpha, 2\beta, 5\alpha, 7\beta)]-3,3,7-Trimethyl-4,9-dioxatricyclo[3.3,1.02.7]nonane; 3,3.7-trimethyl-2,9-di-4,9-dioxatricyclo[3,3,1,0^{3,7}]nonane; 3,3,7-trimethyl-2,9-dioxatricyclo[3,3,1,0^{4,7}]nonane; 4,6,6-lineatin, C₁₀H₁₆O₃; mol wt 168,24. C 71,39%, H 9,59%, O 19,02%. Isoln of the unique tricyclic aggregation pheromone from ambrosia heetles, Trypodendron lineatum (Olivier): J. G. MacConnell et al., J. Chem. Ecol. 3, 549 (1977). Synthesis of (±)-form: K. Mori. M. Sasuki, Tetrahedron Letters 1979, 1329; K. N. Slessor et al., J. Org. Chem. 45, 2290 (1980): K. Mori et al., Tetrahedron Letters 23, 1921 (1982): L. Skattebol, Y. Stenstrom. ibid. 24, 3021 (1983): B. D. Johnston et al., J. Org. Chem. 50, 114 (1985). Synthesis of racemate and optical synthesis. isomers: K. Mori, M. Sasaki, Tetrahedron 36, 2197 (1980). Short stereoselective synthesis: I. Aljancic-Solaja et al., Helv. Chim. Acta 70, 1302 (1987). Comparative activity of the isomers: J. H. Borden et al., Can. Entomol. 112, 107 (1980).

Oil. bp_{10} 70°. $[\alpha]_{D}^{24} + 66.3^{\circ}$ (c = 3.1 in CHCl₃). USF: Insect sex attractant.

Signature of the control of the cont

Colorless oil. Easily oxidized by air, cannot be distilled without decompn. Storage in ester form is recommended. d_1^{18} 0.9038; d_1^{27} 0.9007. mp -12°. bp₁₄ 202°: bp₁₆ 230°. $n_1^{1.5}$ 1.4715; n_1^{20} 1.4699; $n_1^{21.5}$ 1.4683; n_2^{20} 1.4588. lodine value: 181.1. Thiocyanogen value 96.7. Freely sol in ether. Sol in abs alc. One ml dissolves in 10 ml petr ether. Miscible with dimethylformamide, fat solvents, oils.

Aluminum salt, Al(C₁₈H₂₉O₂)₃. Yellow lumps or powder; linseed oil odor. Practically insol in water. Sol in oils, fixed alkali hydroxides.

Methyl Ester see Methyl Linoleate.

Ethyl Ester see Ethyl Linoleate. Cyclohexylamide, C₂H₄₃NO, linolexamide, N-cyclohexyllinoleamide, Clinolamide.

USE: Manuf paints, coatings, emulsifiers, vitamins. Aluminum salt used to manuf lacquers

THERAP CAT: Nutrient (essential fatty acid).

5383. Linolenic Acid. (Z,Z,Z)-9,12,15-Octadecatricnoic acid; a-linolenic acid. C₁₁H₁₉O₂; mol wt 278.42. C 77.65%, H 10.86%, O 11.49%. An essential fatty acid, q.v. Occurs as the glyceride in most drying oils. Synthesis: Nigam, Weedon, J. Chem. Soc. 1956, 4049; Osbond, Wickens, Chem. & Ind. (London) 1959, 1288. Biosynthetic studies: C. G. Kannangara et al., Biochem. Biophys. Res. Commun. 52, 648 (1973); B. S. Jacobson et al., ibid. 1190; C. J. Bedord et al., Arch. Biochem. Biophys. 185, 15 (1978). Effects on lipid metabolism in rat tissue: M. L. Garg et al., Lipids 23, 847 (1988). Review of dietary linolenic acid in mammals: J. Tinoco et al., ibid. 14, 166-171 (1979); in man: N. Zöllner. Prog. Lipid Res. 25, 177-180 (1986). 5383. Linolenic Acid. (Z,Z,Z)-9,12,15-Octadecatricnoic

$$H_1C$$
 (CH₂)₇COOH

Colorless liquid. dis 0.914. bp, 230-232°. Insol in water. THERAP CAT: Nutrient (essential fatty acid).

5384. y-Linolenic Acid. (Z,Z,Z)-6,9,12-Octadecatrienoic acid; cis-6,cis-9,cis-12-octadecatrienoic acid; gamolenic acid; GLA. C₁₈H₃₀O₂; mol wt 278.44. C 77.65%, H 10.86%, O 11.49%. Polyunsaturated fatty acid produced in the body as the Δ^6 -desaturase metabolite of linoleic acid, q.v. Converted to dihomo-y-linolenic acid, a biosynthetic precursor of monoenoic prostaglandins such as PGE₁. Present to varying extents in the fatty acid fraction of evening primrose oil (7-10%), in borage oil (18-26%), in black currant oil (15-20%) and in oils from different fungal sources (6-24%). 20%) and in oils from different tungal sources (6-24%). Isoln from evening primrose oil, q.v.: A. Heiduschka, K. Luft, Arch. Pharm. 257, 33 (1919). Proposed structure: Eibner et al., Chem. Umschau. 34, 312 (1927). Confirmation of structure: J. P. Riley, J. Chem. Soc. 1949, 2728. Discussion of occurrence, esp. in fungi: R. Shaw, Biochim. Biophys. Acta 98, 230 (1965). Synthesis: J. M. Osbond et al., J. Chem. Soc. 1961, 2779; J. M. Osbond. ibid. 5270. Metabolism studies: J. F. Mead, D. R. Howton, J. Biol. Chem. 229, 575 (1957); K. J. Stone et al., Linids 14, 174 (1979). Effect 18m studies: J. F. Meau, D. K. Holdon, J. Land St. 1575 (1957); K. J. Stone et al., Lipids 14, 174 (1979). Effect of source on essential fatty acid and prostanoid metabolite formation: D. K. Jenkins et al., Med. Sci. Res. 16, 525

Hexabromide deriv. $C_{18}H_{30}Br_4O_7$, crystals from ethyl methyl ketone, mp 201-202°. USE: Nutrient.

THERAP CAT: In treatment of atopic eczema.

5385. Linseed, Flaxseed; linum. Dried ripe seeds of num usitatissimum L., Linaceae. Source of linseed oil. Linum usitatissimum L., Linaceae. Source of linseed oil. Constit. 30-40% oil, about 6% mucilage, about 25% proteins and linamarin.

USE: Emollient.
THERAP CAT (VET): Poultice (crushed seeds), demulcent (boiled in water).

5386. Linseed Oil, A drying oil obtained by expression of linseed. Constit. Glycerides of linolenic, linoleic, oleic, stearic, palmitic and myristic acids. Ref: T. P. Hilditch. The Chemical Constitution of Natural Fats (London. 3rd ed..